Synthesis of fluoro-substituted 4,5-dihydro-1,2,4oxadiazoles via 1,3-dipolar cycloadditions[†]

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4,5-Dihydro-1,2,4-oxadiazoles have been prepared by regiospecific 1,3-dipolar cycloaddition of nitrile oxides with fluoro-substituted aldimines.

Keywords: cycloaddition, fluoro compounds, 1,2,4-oxadiazoles, imines, nitrile oxides

The importance of fluoro-heterocycles is widely acknowledged in many fields, such as medicine and agrochemistry, due to their peculiar properties and biological activity.¹ Among them, many fluoro-substituted, fully unsaturated 1,2,4-oxadiazoles are described in the literature, owing to their interest as pesticides and insecticides² or in medicinal chemistry³, e.g. as antiviral agents. Nevertheless, only very few examples of the corresponding fluoro-substituted partially hydrogenated 4,5-dihydro-1,2,4-oxadiazoles are known.⁴

For the construction of heterocyclic systems several methods are available: among these, 1,3-dipolar cycloadditions occupy a preeminent place, owing to their versatility and to the high regio- and stereo-chemical control that can be involved in such reactions.5

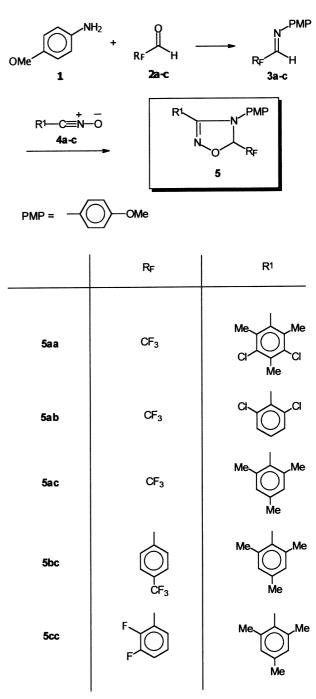
Within a research program aimed to the synthesis of fluorinated heterocycles of biological interest via 1,3-dipolar cycloadditions, 6 we thought to devise a cycloadditive route to 4,5-dihydro-1,2,4-oxadiazoles, starting from common 1,3dipoles and easily accessible fluorinated dipolarophiles. The present paper describes the synthesis of 5-trifluoromethyl- and 5-(fluoroaryl)-4,5-dihydro-1,2,4-oxadiazoles through the cyclo-addition of the aromatic nitrile oxides 4 with fluorosubstituted aldimines 3 (Scheme 1).

Results and discussion

The synthesis of fluoro-substituted aldimines 3a-c was performed in good yield (75-90%) through the condensation of 4-methoxybenzenamine (1) with fluoroaldehydes 2a-c in the presence of an acidic ion-exchange resin (Dowex W X 8-400). Subsequently, the nitrile oxides **4a–c** were treated with a 1.2:1 molar excess of the fluoro-substitued aldimines 3a-c in a CCl₄ solution: after 3 to 10 days at room temperature, the cycloadducts 5 were isolated in moderate to good yields as the only regioisomers. The assigned structures rely upon analytical and spectral data (Table 1): in particular the ¹³C NMR signals of the sp^3 ring carbon atom, ranging from 91 to 97 ppm (see Experimental), are evidence of its position between an oxygen and a nitrogen atom; the observed regiochemical orientation, that can be ascribed mainly to electronic factors, is also in agreement with literature data on the 1,3-dipolar cycloadditions of nitrile oxides with C=N bonds.⁷

Owing both to the great variety of fluorinated aldimines that can be prepared by the above described method, and to the number of available nitrile oxides, the reported reactions can constitute an effective and versatile cycloadditive route to a

[†] This is a Short Paper, there is therefore no corresponding material in J Chem. Research (M).



Scheme 1

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Product	δ_{H} (ppm), J (Hz)	δ_{F} (ppm), J (Hz)
5aa	7.02 and 6.74 (m, 4 H, ArH), 6.18 (q, 1 H, ³ J _{H,F} 4.2, CHN), 3.72 (s, 3 H, OMe), 2.51, 2.46 and 2.26 (s, 9 H, 3xAr <i>Me</i>)	–83.52 (d, 3 F, ³ J _{EH} 4.2, CF ₃)
5ab	7.4-7.1 and 6.78 (m, 7 H, ArH), 6.08 (q, 1 H, ³ J _{HE} 4.1, CHN), 3.74 (s, 3 H, OMe)	–83.56 (d, 3 F, ³ J _{EH} 4.1. CF₃)
5ac	7.08, 6.92, 6.82 and 6.80 (m, 6 H, ArH), 6.81 (q, 1 H, ³ J _{H,F} 4.6, CHN), 3.68 (s, 3 H, OMe), 2.36, 2.20 and 2.08 (br s. 9 H, 3×Ar <i>Me</i>)	–82.91 (d, 3 F, ³ J _{F,H} 4.6, CF ₃)
5bc	7.70, 7.67, 6.80, 6.65 and 6.60 (m, 10 H, ArH), 6.88 (s, 1 H, CHN), 3.63 (s, 3 H, OMe), 2.45, 2.24 and 2.20 (br s, 9 H, 3×Ar <i>Me</i>)	–63.2 (br s, 3 F, CF ₃)
5cc	7.6–6.5 (m, 9 H, ArH), 7.23 (s, 1 H, CHN), 3.63 (s, 3 H, OMe), 2.49, 2.24 and 2.15 (s, 9 H, 3×Ar <i>Me</i>)	–138.9 and –145.8 (m, 2 F, 2×ArF)

Table 1 ¹H and ¹⁹F NMR data of 4,5-dihydro-1,2,4-oxadiazoles 5

class of fluoro-substituted heterocycles of biological interest. As a further development of this work, we are now considering the use as dipolarophiles of fluoro-substituted aldimines bearing also a chiral and enantiomerically pure sulfinyl function. The asymmetric induction that this chiral auxiliary is able to exert in the cycloadditions could allow the preparation of enantiopure fluoro-substituted 4,5-dihydro-1,2,4-oxadiazoles. These reactions are in progress and will be reported in due course.

Experimental

Melting points were obtained using a capillary apparatus and are uncorrected. Analytical TLCs were performed with Merck silica gel 60 F₂₅₄ plates. Flash column chromatographies were performed with silica gel 60 (230-400 ASTM mesh). ¹H, ¹³C and ¹⁹F NMR spectra were obtained on a Bruker AC 250L spectrometer, operating at 250.13, 62.89 and 235.35 MHz, respectively, in CDCl₃ solutions. Chemical shifts are expressed in ppm (δ) , using tetramethylsilane (TMS) as internal standard for ¹H and ¹³C nuclei ($\delta_{\rm H}$ and $\delta_{\rm C}$ = 0.00), whilst C_6F_6 was used as external standard ($\delta_F = -162.90$) for ¹⁹F. Coupling constants are expressed in Hertz (Hz). In the ¹³C NMR signal assignment, capital letters refer to the pattern resulting from directly bonded (C,H) couplings and lower case letters to that from (C,F) couplings. Mass spectra were registered on a TSQ 70 Finnigan Mat three-stage quadrupole instrument. DIS (Direct Inlet System) was used for pure compounds. Infrared spectra were obtained using a Perkin-Elmer System 2000 FT-IR.

Commercially available reagents and solvents were employed without further purification. Trifluoroacetaldehyde **2a** was directly used in the commercial hydrate form. Nitrile oxides **4a**¹⁰ and **4c**¹¹ were prepared by standard literature methods. Nitrile oxide **4b** was generated *in situ* by treatment of the corresponding chlorooxime¹² with an excess of triethylamine.

Fluoro-substituted aldimines 3. General procedure: To a solution of 4-methoxybenzenamine 1 (0.96 g, 7.75 mmol) in benzene (9.4 ml) fluoroaldehydes 2a-c (7.04 mmol) were added, followed by a catalytic amount of Dowex 50 W X 8-400 ion-exchange resin. The mixture was refluxed for 2 hours, collecting the water of condensation by a Dean-Stark apparatus. After filtering off the catalyst and evaporating the solvent under reduced pressure, the residue was distilled in vacuo (3a) or flash chromatographed on a silica gel column (3b,c) with hexane-ethyl acetate mixtures, affording fluoro-substituted aldimines 3a-c, in yields ranging from 75 to 90%. Physical properties of the aldimines 3a¹³ and 3b¹⁴ were consistent with literature data.

N-(*4-Methoxyphenyl*)-2,*3-difluorobenzaldimine* **3c:** yield 88%; m.p. 53–54°C; v_{max} (KBr) 1623 cm⁻¹. $\delta_{\rm H}$ 8.78 (s, 1 H, CH=N); 7.93 (dddd, 1 H, *J* 7.9, 5.9, 1.9, 1.7), 7.22 (dddd, 1 H, *J* 9.6, 7.8, 7.5, 1,9) and 7.16 (dddd, 1 H, *J* 7.9, 7.8, 5.0, 1.6): H-4, H-5 and H-6; 7.27 and 6.94 (m, 4 H, ArOMe*H*); 3.85 (s, 3 H, OMe). $\delta_{\rm F}$ -139.75 (dddd, 1 F, *J* 20.0, 9.6, 5.0, 1.7) and -147.86 (dddd, 1 F, *J* 20.0, 7.5, 5.9, 1.6): F-2 and F-3. $\delta_{\rm C}$ 155.88 (S, C-4'); 150.73 (Sdd, $J_{\rm C,F}$ 256 and 13) and 150.63 (Sdd, $J_{\rm C,F}$ 249 and 12): C-2 and C-3; 149.83 (Dt, $J_{\rm C,F}$ 4.5, CH=N); 144.24 (S, C-1'); 126.36 (Sbrd, $J_{\rm C,F}$ 7.5, C-1); 124.24 (Ddd,

 $J_{\rm C,F}$ 7 and 5), 122.39 (Dm) and 119.24 (Dbrd, $J_{\rm C,F}$ 17.5): C-4, C-5 and C-6; 122.45 and 114.45 (D, ArOMe*CH*); 55.52 (Q, OMe). m/z (%) 247 (M+, 100), 232 (70), 134 (5), 107 (8), 77 (18). (Found: C, 66,23; H, 4.59; N, 6.06. $C_{14}H_{11}F_2NO$ requires: C, 66.36; H, 4.72; N, 5.96).

4,5-Dihydro-1,2,4-oxadiazoles 5. General procedure: A solution of nitrile oxide 4 (0.50 mmol) and aldimine 3 (0.60 mmol) in CCl₄ (5.0 ml) was stirred at room temperature for a time ranging from 2 to 9 days. After removal of the solvent under reduced pressure, the residue was flash chromatographed on a silica gel column with hexane-ethyl acetate mixtures, affording 4,5-dihydro-1,2,4-oxadiazoles 5 (see Scheme 1 and Table 1).

 $3\text{-}(3,5\text{-}Dichloro\text{-}2,4,6\text{-}trimethylphenyl})\text{-}4\text{-}(4\text{-}methoxyphenyl})\text{-}5\text{-}trifluoromethyl\text{-}4,5\text{-}dihydro\text{-}1,2,4\text{-}oxadiazole}$ **5aa:** reaction time 3 days; yield 34%; m.p. 134–135°C. $\delta_{\rm C}$ 159.2 (S), 154.5 (S), 137.2 (S), 134.9 (2×S), 133.8 (2×S), 128.7 (S), 127.5 (2×D), 122.3 (S), 121.6 (Sq, $^1J_{\rm C,F}$ 287.5), 114.9 (2×D), 91.9 (Dq, $^2J_{\rm C,F}$ 35.5), 55.4 (Q), 19.0 (Q), 18.8 (Q), 18.3 (Q). m/z (%) 434 (40), 432 (M+, 60), 363 (43), 335 (30), 134 (68), 122 (100), 108 (38). (Found: C, 52.49; H, 4.04; N, 6.32. $C_{19}H_{17}Cl_2F_3N_2O_2$ requires: C, 52.67; H, 3.95; N, 6.47).

 $3\text{-}(2,6\text{-}Dichlorophenyl)\text{-}4\text{-}(4\text{-}methoxyphenyl)\text{-}5\text{-}trifluoromethyl}\text{-}4,5\text{-}dihydro\text{-}1,2,4\text{-}oxadiazole}$ **5ab:** reaction time 4 days; yield 67%; m.p. 123–124°C. $\delta_{\rm C}$ 159.4 (S), 152.2 (S), 136.3 (S), 136.2 (S), 132.4 (2×D), 128.5 (2×D), 128.4 (D), 128.2 (S), 122.6 (S), 121.6 (Sq, $^1\!J_{\rm C,F}$ 293.0), 114.7 (2×D), 92.4 (Dq, $^2\!J_{\rm C,F}$ = 35.5), 55.4 (Q). m/z (%) 392 (65), 390 (M*, 100), 321 (92), 293 (94), 134 (37), 122 (80), 108 (60) (Found: C, 49.01; H, 2.95; N, 7.03. $C_{16}H_{11}Cl_2F_3N_2O_2$ requires: C, 49.13; H, 2.83; N, 7.16).

 $4\text{-}(4\text{-}Methoxyphenyl)\text{-}5\text{-}trifluoromethyl\text{-}3\text{-}(2,4,6\text{-}trimethylphenyl)\text{-}}4,5\text{-}dihydro\text{-}1,2,4\text{-}oxadiazole}$ **5ac:** reaction time 2 days; yield 81%; oil. δ_{C} 158.6 (S), 154.9 (S), 140.4 (S), 138.2 (2×S), 130.1 (S), 128.8 (2×D), 126.7 (2×D), 122.0 (Sq, $^{1}J_{\mathrm{C,F}}$ 288.5), 119.4 (S), 114.7 (2×D), 91.6 (D, $^{2}J_{\mathrm{C,F}}$ 35.5), 55.3 (Q), 21.1 (Q), 20.0 (Q), 19.6 (Q). m/z (%) 365 (18), 364 (M+, 100), 295 (60), 267 (10), 134 (40), 122 (24) (Found: C, 62.85; H, 5.08; N, 7.49. $C_{19}H_{19}F_{3}N_{2}O_{2}$ requires: C, 62.63; H, 5.26; N, 7.69).

4-(4-Methoxyphenyl)-5-(4-trifluoromethylphenyl)-3-(2,4,6-trimethylphenyl)-4,5-dihydro-1,2,4-oxadiazole **5bc:** reaction time 7 days; yield 27%; oil. $\delta_{\rm C}$ 157.2 (S), 154.3 (S), 141.7 (S), 140.0 (S), 137.8 (2×S), 132.1 (Sq., $^2J_{\rm CF}$ 33), 129.9 (S), 128.6 (2×D), 128.2 (Sq. $^1J_{\rm CF}$ 275), 128.1 (2×D), 125.7 (2×Dq, $^3J_{\rm CF}$ 5), 124.3 (2×D), 120.9 (S), 114.4 (2×D), 96.7 (D), 55.3 (Q), 21.2 (Q), 20.0 (2×Q). m/z (%) 440 (M⁺, 30), 279 (38), 266 (100), 265 (32), 134 (68), 251 (21) (Found: C, 67.96; H, 5.40; N, 6.48. C₂₅H₂₃F₃N₂O₂ requires: C, 68.17; H, 5.26; N, 6.36).

 $5\cdot(2\,,3\text{-}Difluorophenyl)\text{-}4\text{-}(4\text{-}methoxyphenyl)\text{-}3\text{-}(2\,,4\,,6\text{-}trimethylphenyl)\text{-}4,5\text{-}dihydro\text{-}1,2,4\text{-}oxadiazole}$ **5cc:** reaction time 12 days; yield 31%; oil. $\delta_{\rm C}$ 157.1 (S), 154.0 (S), 150.4 (Sm), 149.5 (Sm), 140.0 (S), 137.9 (2×S), 129.7 (S), 128.6 (2×D), 127.5 (Sm), 124.6 (Dm), 124.5 (Dm), 123.9 (2×D), 120.9 (S), 118.5 (Dm), 114.3 (2×D), 91.2 (D), 55.2 (Q), 21.2 (Q), 19.9 (2×Q). m/z (%) 408 (M+, 38), 266 (90), 265 (100), 247 (80), 232 (70), 161 (18), 147 (37), 141 (43), 130 (25), 92 (15), 77 (10) (Found: C, 70.34; H, 5.31; N, 6.73. $C_{24}H_{22}F_{2}N_{2}O_{2}$ requires: C, 70.58; H, 5.43; N, 6.86).

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References

- (a) Biomedical Aspects of Fluorine Chemistry, eds. R. Filler and A. Kobayashi, Elsevier, New York, 1992; (b) M.J. Silvester, Adv. Heterocycl. Chem., 1994, 59, 1; (c) K. Burger, U. Wucherpfennig and E. Brunner, Adv. Heterocycl. Chem., 1994, 60, 1: (d) Fluoroorganic Chemistry: Synthetic Challenges and Biomedical Rewards, eds. G. Resnati and V.A. Soloshonok, Tetrahedron Symposium-in-Print 58, Tetrahedron, 1996, 52, 1; (e) Asymmetric Fluoroorganic Chemistry: Synthesis, Applications, and Future Directions, ed. P.V. Ramachandran, ACS Symposium Series 746, ACS, Washington D.C., 2000.
- (a) Y. Ozoe, K. Yagi, M. Nakamura, M. Akamatsu, T. Miyake and F. Matsumura, *Pestic. Biochem. Physiol.*, 2000, **66**, 92; (b) N.V. Kirby, E.J. Canada, I.M. Morrison, M.E. Pieczko, G.D. Gustafson, J.T. Mathieson, D.H. Cooper, C.S. Galka and J.L. Adamski, US Patent 2000015637, 2000, Chem. Abstr., 2000, 132, 222538; (c) T. Kishimoto, K. Noda, Y. Shibata, M. Matsuda, R. Hatano, M. Yano and T. Iwasa, PCT Int. Appl. WO 94 17048, Chem. Abstr., 1994, 121, 280650d.
- (a) C. Burkholder, W.R. Dolbier Jr. and M. Médebielle, J. Fluorine Chem., 1999, 95, 127; (b) C. Burkholder, W.R. Dolbier Jr. and M. Médebielle, J. Org. Chem., 1998, 63, 5385; (c) D.C. Pevear, T.M. Tull, M.E. Seipel and J.M. Groarke, Antimicrob. Agents Chemother., 1999, **43**, 2109; (d) D.N. Nicolaides, K.C. Fylaktakidou, K.E. Litinas and D. Hadjipavlou-Litina, Eur. J. Med. Chem., 1998, 33, 715; (e) G.D. Diana, P. Rudewicz, D.C. Pevear, T.J. Nitz, S.C. Aldous, D.J. Aldous, D.T. Robinson, T. Draper and F.J. Dutko, *J. Med.* Chem., 1995, 38, 1355; (f) J.B. Hynes and R.F. Gratz, J. Med. Chem., 1972, 15, 1198.
- 4 (a) L.A. Simonyan., Yu.V. Zeifman and N.P. Gambaryan, Izv. Akad. Nauk SSSR, Ser. Khim., 1968, 8, 1916, Chem. Abstr., 1969, 70, 3964y; (b) N.V. Vasil'ev, A.F. Kolomiets and G.A. Sokol'skii, Zh. Vses. Khim. O-va., 1980, 25, 703, Chem. Abstr. 1981, 94, 174991; (c) V.A. Soloshonok and V.P. Kukhar, Zh. Org. Khim., 1990, 26, 419, Chem. Abstr., 1990, 113, 59045.
- (a) 1,3-Dipolar Cycloaddition Chemistry, ed. A. Padwa, Wiley, New York, 1984; (b) W. Carruthers, Cycloaddition Reactions in Organic Synthesis, Pergamon Press, Oxford, 1990; (c)

- K.B.G. Torssell. Nitrile Oxides Nitrones and Nitronates in Organic Synthesis, VCH, New York, 1988; (d) M. Cinquini and F. Cozzi, in Houben-Weyl, 4th ed., vol. E21c, eds. G. Helmchen, R.W. Hoffmann, J. Mulzer and E. Schaumann, Thieme, Stuttgart, 1995, p. 2953; (e) M. Frederickson, Tetrahedron, 1997, 53, 403; (f) K.V. Gothelf and K.A. Jorgensen, Chem. Rev., 1998, 98, 863.
- (a) P. Bravo, L. Bruché, A. Farina, G. Fronza, S.V. Meille and A. Merli, Tetrahedron: Asymmetry, 1993, 4, 2131; (b) A. Arnone, P. Bandiera, P. Bravo, L. Bruché and M. Zanda, Gazz, Chim. Ital., 1996, 126, 773; (c) A. Arnone, P. Bravo, L. Bruché, P. Seresini and M. Zanda, J. Heterocycl. Chem., 1997, 34, 489; (d) A. Arnone, P. Bravo, L. Bruché, W. Panzeri, C. Pesenti and F. Viani, Eur. J. Org. Chem., 1999, 1665.
- (a) L.B. Clapp, in Comprehensive Heterocyclic Chemistry, vol. 6, ed. K.T. Potts, Pergamon Press, Oxford, 1984, p. 389. (b) J.C. Jochims, in Comprehensive Heterocyclic Chemistry II, eds. A.R. Katritzky, C.W. Rees and E.F.V. Scriven, Pergamon Press, Oxford, 1996, vol. 4, p. 215. (c) B. Alcaide, C.L. Mardomingo, J. Plumet, C. Cativiela and J.A. Mayoral, Can. J. Chem., 1987, 65, 2050. (d) G. Grassi, F. Risitano and F. Foti, Tetrahedron, 1995, 51, 11855. (e) D. Enders, I. Meyer, J. Runsink and G. Raabe, Heterocycles, 1999, **50**, 995.
- 8 P. Bravo, M. Crucianelli, B. Vergani and M. Zanda, Tetrahedron Lett., 1998, 39, 7771.
- 9 (a) T. Takahashi, A. Fujii, J. Sugita, T. Hagi, K. Kitano, Y. Arai, T. Koizumi and M. Shiro, Tetrahedron: Asymmetry, 1994, 5, 987. (b) P. Bravo, L. Bruché, M. Crucianelli, A. Farina, S.V. Meille, A. Merli and P. Seresini, J. Chem. Res., 1996, (S) 348, (M) 1901.
- 10 P. Beltrame, C. Veglio and M. Simonetta, J. Chem. Soc. B, 1967,
- 11 K.-C. Liu, B.R. Shelton and R.K. Howe, J. Org. Chem., 1980, 45, 3916.
- 12 J.N. Kim, K.S. Jung, H.J. Lee and J.S. Son, Tetrahedron Lett., 1997, 38, 1597.
- 13 P. Bravo, A. Farina, V.P. Kukhar, A.L. Markovsky, S.V. Meille, V.A. Soloshonok, A.E. Sorochinsky, F. Viani, M. Zanda and C. Zappalà, J.Org.Chem., 1997, 62, 3424.
- I. Ojima, I. Habus, M. Zhao, M. Zucco, Y.H. Park, C.M. Sun and T. Brigaud, Tetrahedron, 1992, 48, 6985.